



# A new general approach to 4-substituted-3-halo-2-quinolones

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## ABSTRACT

A general procedure for the preparation of 4-substituted-3-halo-2-quinolones (halo = F, Cl, Br) utilizing 2-halo diethylphosphonoacetic acids (halo = F, Cl, Br) and *o*-aminophenylketones as the starting materials is described. The title compounds are obtained by an intramolecular Horner–Wadsworth–Emmons olefination of halogen-containing *N*-acyl-*o*-aminophenylketones. The transformation process is generally applicable under mild conditions.

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## 1. Introduction

The 2-quinolone skeleton, as an important structure unit, usually presents in a large number of alkaloids and in biologically active compounds [1–3]. There has been considerable interest in developing 2-quinolones as anticancer [4], antibacterial [5], antihypertensive agents [6], and inhibitors of the macrophage colony-stimulating factor-1 receptor (FMS) and p38 Mitogen-activated protein (MAP) kinase [7,8]. Moreover, 2-quinolones are useful precursors to the corresponding quinolines and isoquinolines [9].

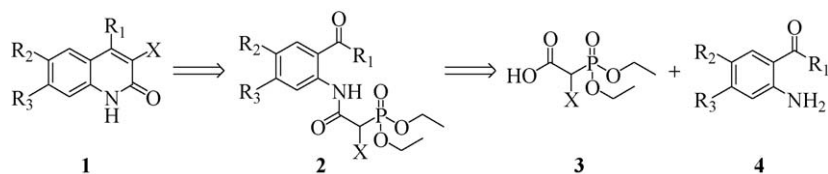
In recent years, an increasing interest in the synthesis of functionalized 2-quinolones with promising biological properties has been observed [10]. Especially, 4-substituted-3-halo-2-quinolones have been given more attention, which have been used as cardiac stimulants [11] and herbicides [12]. However, to the best of our knowledge, there is no general method for the synthesis of 4-substituted-3-halo-2-quinolones (halo = F, Cl, Br) reported. Some examples used direct halogenation of 4-substituted-2-quinolones, such as the bromination of 4-substituted-2-quinolones with NBS or bromine dissolved in acetic acid to prepare 4-substituted-3-bromo-2-quinolones [13]. Some used substitution of 4-substituted-2-quinolone derivatives, such as Rh<sub>2</sub>(OAc)<sub>4</sub> catalyzed cyclic diazodicarbonyl compounds with a variety of halides obtaining the corresponding 4-substituted-3-chloro(or 3-bromo)-2-quinolones [14]. And the substitution of 3-bromo-4-trifluoro-quinolin-2(1*H*)-one with Pd(Cl)<sub>2</sub>(PPh)<sub>3</sub> was used to get 4-substituted-3-bromo-2-quinolones [15]. A frequently employed method is an intramolec-

ular condensation of an *N*-acyl-*o*-aminophenylketone. For example, the condensation of  $\alpha$ -chloro(or bromo)-aroylacetylides with concentrated sulfuric acid giving the corresponding 4-substituted-3-chloro(or bromo)-2-quinolones [16,17], and the condensation of 2-chloro-2-(2-fluorobenzoyl) acetanilide with 85% KOH in DMF and H<sub>2</sub>O giving 3-chloro-4-(2-fluorophenyl)-2(1*H*)-quinolinone [18]. Nevertheless, the limitation of methods reported lies in the difficulty of halogenation due to the difficulty in controlling their regioselectivity, the strong reaction conditions, and the side reactions involving polyhalogenation [19]. Other disadvantages involve the low yields of the reaction, a cumbersome and lengthy process, migration of substituents (rearrangement), and not readily available starting materials. Therefore, it is required to develop a general, effective and simple method to prepare 4-substituted-3-halo-2-quinolones. To pursue the goal, we adopted the strategy shown in Scheme 1. In order to obtain compounds **1** with high yields and easily controlled regioselectivity of reactions, an intramolecular Horner–Wadsworth–Emmons olefination of intermediates **2** was employed. The phosphonoacetate in molecules of intermediates **2** activated the carboxamide  $\alpha$ -position, which could be eliminated after cyclization under basic conditions to produce compounds **1**. In this strategy, halogens were previously introduced into intermediates **2** by the acylation of 2-halo diethylphosphonoacetic acids **3** with *o*-aminophenylketones **4**, which could overcome the difficulty in controlling their regioselectivity.

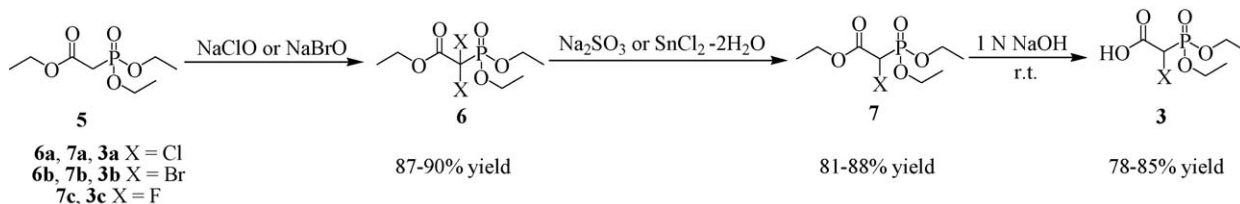
## 2. Results and discussion

One of the starting materials 2-halo diethylphosphonoacetic acids **3** were easily prepared in good yields by the hydrolysis of triethyl 2-

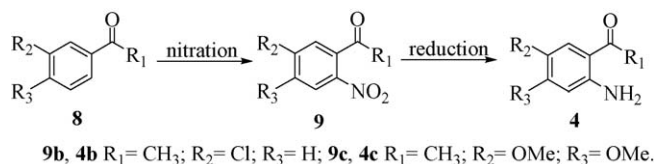
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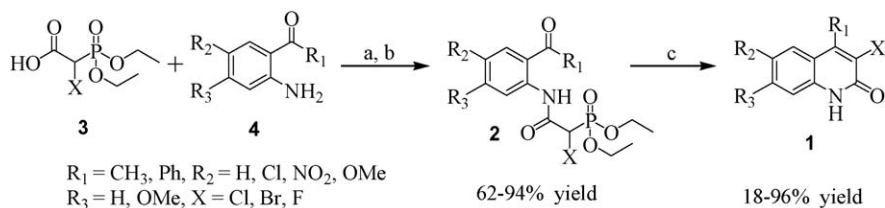
**Scheme 1.** Strategy for the preparation of 4-substituted-3-halo-2-quinolones.



**Scheme 2.** Preparation of 2-halo diethylphosphonoacetic acids **3**.



**Scheme 3.** Preparation of *o*-aminophenylketones.



**Scheme 4.** Preparation of 4-substituted-3-halo-2-quinolones. Reagents and conditions: (a) (COCl)<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, r.t., 24 h; (b) pyridine, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to r.t., 4.5 h; (c) THF, LiCl, DBU, r.t., 2 h.

halo phosphonoacetates **7** with 1 N NaOH [20]. Triethyl 2-fluorophosphonoacetate **7c** was commercially available. However, triethyl 2-chlorophosphonoacetate **7a** and triethyl 2-bromophosphonoacetate **7b** were prepared according to literatures [21]. The method is illustrated in **Scheme 2**. Triethyl 2,2-dichlorophosphonoacetate **6a** was prepared by the chlorination of triethyl phosphonoacetate **5** using sodium hypochlorite, followed by selective reduction to convert the dichloro product to the corresponding triethyl 2-chlorophosphonoacetate **7a** with sodium sulfite. In the same way, triethyl 2,2-dibromophosphonoacetate **6b** was prepared by the bromination of triethyl phosphonoacetate **5** using the freshly prepared sodium hypobromite, followed by selective reduction to convert the dibromo product to the corresponding triethyl 2-bromophosphonoacetate **7b** with SnCl<sub>2</sub>·2H<sub>2</sub>O.

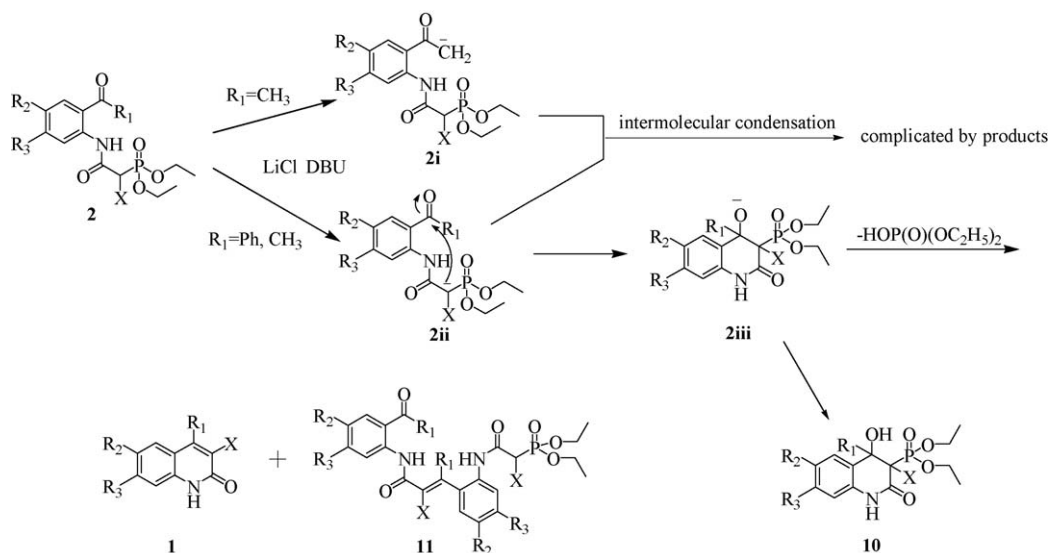
Another starting materials **4a**, **4d–f** were purchased from commercial sources. However, 2-amino-5-chloroacetophenone **4b** and 2-amino-4,5-dimethoxyacetophenone **4c** were prepared by ourselves. The method is illustrated in **Scheme 3**. 3-Chloroacetophenone **8b** was converted to 5-chloro-2-nitroacetophenone **9b** by nitration with fuming HNO<sub>3</sub> and concentrated H<sub>2</sub>SO<sub>4</sub> at –20 °C, which was then reduced by iron powder in AcOH to form 2-amino-5-chloroacetophenone **4b** [22]. 3,4-Dimethoxyacetophenone **8c** was converted to 4,5-dimethoxy-2-nitroacetophenone **9c** by nitration with fuming HNO<sub>3</sub> and AcOH at 0 °C, which was then reduced into 2-amino-4,5-dimethoxyacetophenone **4c** by H<sub>2</sub> with PtO<sub>2</sub>/C [23].

The intermediates **2** were prepared by the acylation of *o*-aminophenylketones **4a–f** with 2-halo diethylphosphonoacetic acids **3a–c** [8]. The method is illustrated in **Scheme 4**. In order to realize this transformation, **3** were first converted into acyl

**Table 1**

The preparation of intermediates **2a–r** using 2-halo diethylphosphonoacetic acids **3** and *o*-aminophenylketones **4**.

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	X	Conversion (%)	Yield (%)
<b>2a</b>	CH <sub>3</sub>	H	H	Cl	98	94
<b>2b</b>	CH <sub>3</sub>	Cl	H	Cl	96	93
<b>2c</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	Cl	95	92
<b>2d</b>	Ph	H	H	Cl	74	70
<b>2e</b>	Ph	Cl	H	Cl	95	90
<b>2f</b>	Ph	NO <sub>2</sub>	H	Cl	92	88
<b>2g</b>	CH <sub>3</sub>	H	H	Br	95	91
<b>2h</b>	CH <sub>3</sub>	Cl	H	Br	96	92
<b>2i</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	Br	94	91
<b>2j</b>	Ph	H	H	Br	69	65
<b>2k</b>	Ph	Cl	H	Br	91	87
<b>2l</b>	Ph	NO <sub>2</sub>	H	Br	90	85
<b>2m</b>	CH <sub>3</sub>	H	H	F	98	92
<b>2n</b>	CH <sub>3</sub>	Cl	H	F	98	92
<b>2o</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	F	96	90
<b>2p</b>	Ph	H	H	F	65	62
<b>2q</b>	Ph	Cl	H	F	94	89
<b>2r</b>	Ph	NO <sub>2</sub>	H	F	93	88



**Scheme 5.** Proposed mechanism for the reaction of N-acyl-*o*-aminophenylketones with LiCl and DBU.

**Table 2**

The yields of 4-substituted-3-halo-2-quinolones **1** via a Horner–Wadsworth–Emmons route.

Entry	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	X	Purification methodology	mp (°C)	Yield (%)
<b>1a</b>	CH <sub>3</sub>	H	H	Cl	Recrystallization	300–302	88
<b>1b</b>	CH <sub>3</sub>	Cl	H	Cl	Recrystallization	272–274	95
<b>1c</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	Cl	Recrystallization	250–252	82
<b>1d</b>	Ph	H	H	Cl	Recrystallization	191–193	94
<b>1e</b>	Ph	Cl	H	Cl	Recrystallization	299–301	95
<b>1f</b>	Ph	NO <sub>2</sub>	H	Cl	Recrystallization	221–223	96
<b>1g</b>	CH <sub>3</sub>	H	H	Br	Recrystallization	268–269	85
<b>1h</b>	CH <sub>3</sub>	Cl	H	Br	Recrystallization	204–206	88
<b>1i</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	Br	Recrystallization	227–229	78
<b>1j</b>	Ph	H	H	Br	Column chromatography	228–230	26
<b>1k</b>	Ph	Cl	H	Br	Column chromatography	218–220	18
<b>1l</b>	Ph	NO <sub>2</sub>	H	Br	Column chromatography	202–204	19
<b>1m</b>	CH <sub>3</sub>	H	H	F	Column chromatography	179–181	40
<b>1n</b>	CH <sub>3</sub>	Cl	H	F	Column chromatography	211–213	38
<b>1o</b>	CH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	F	Column chromatography	266–268	30
<b>1p</b>	Ph	H	H	F	Recrystallization	254–256	91
<b>1q</b>	Ph	Cl	H	F	Recrystallization	197–199	92
<b>1r</b>	Ph	NO <sub>2</sub>	H	F	Recrystallization	188–190	93

halide *in situ* by using (COCl)<sub>2</sub> at room temperature, which then reacted with newly added *o*-aminophenylketones **4** in the presence of excessive pyridine to give compounds **2a–r**. The reaction underwent smoothly, and good yields were obtained except 2-aminodiphenyl methanone **4d** as substrate. The results are shown in Table 1. As can be seen from Table 1, the reaction was applicable to a variety of *o*-aminophenylketones **4**. The successful acylation of **4** with **3** was achieved when R<sub>1</sub> = alkyl and aryl, and R<sub>2</sub>, R<sub>3</sub> as electron-donating and electron-withdrawing groups. In order to test the generality of this method, we embarked on a systematic study of substituted *o*-aminophenylketones **4** and 2-halo diethylphosphonoacetic acids **3** as starting materials.

Compounds **2** were subjected to cyclization under the mildly basic Masamune–Roush conditions (LiCl/DBU) [24]. As it turned out, the difference of the halogen and substituents had effect on the cyclization of the intermediates **2**. When X = Cl, all the cyclization of intermediates **2a–f** occurred smoothly to produce the expected 2-quinolones **1a–f** in excellent yields (Entries **1a–f**, Table 2).

However, in the case of X = Br, when R<sub>1</sub> = alkyl, the cyclization of intermediates **2g–i** gave the expected 2-quinolones **1g–i** in good yields, but when R<sub>1</sub> = aryl, the reactions were complicated, and the

cyclization of intermediates **2j–l** gave **1j–l** only in yields of 18–26%. The expected products were obtained only as the minor products, and the unexpected byproducts **10** were isolated as the main one. Interestingly, the contrast results were observed for the cyclization of intermediates **2** with X = F. When R<sub>1</sub> = aryl, the cyclization of intermediates **2p–r** occurred smoothly to produce the expected 2-quinolones **1p–r** in excellent yields, but when R<sub>1</sub> = alkyl, the cyclization of intermediates **2m–o** gave **1m–o** only in yields of 30–40%, and a few byproducts were observed.

Based on the above results, we proposed the possible mechanism of the cyclization (Scheme 5). Two carbanions **2i** and **2ii** could be formed under basic conditions. Some unexpected products could be formed via carbanions **2i** by inter- and intramolecular condensations. Not only can carbanions **2ii** give expected products **1** and uneliminated products **10**, but also it can form unexpected products **11** by intermolecular condensation.

### 3. Conclusions

The reaction of *o*-aminophenylketones with 2-halo diethylphosphonoacetic acids provides a general procedure for 4-substituted-3-halo-2-quinolones from readily available starting materials. The

cyclization of halogenated N-acyl-o-aminophenylketones is mainly influenced by the halogen. Although some compounds were obtained in low yields, this new method may be considered as new halogen-containing substrates for the synthesis of a wide variety of heterocyclic compounds with potential biological activity. Study of this reaction and the nature of the products are being continued.

#### 4. Experimental

Infrared spectra were measured on a Bruker Tensor 27 IR spectrophotometer.  $^1\text{H}$  NMR (300 MHz),  $^{13}\text{C}$  NMR (75 MHz),  $^{31}\text{P}$  NMR (121 MHz) and  $^{19}\text{F}$  NMR (282 MHz) spectra were recorded on a Bruker AV-300. The spectra were recorded in  $\text{CDCl}_3$  and  $\text{DMSO}-d_6$  as solvent at room temperature. The chemical shifts ( $\delta$ ) were given in ppm, and the coupling constants ( $J$ ) in Hz. HRMS spectra were recorded on a high resolution ESI-FTICR and MALDI-FTICR mass spectrometry. Melting points were determined on a X-4 digital display micromelting point apparatus and were uncorrected.

##### 4.1. Triethyl 2-chlorophosphonoacetate (7a) [25]

A solution of 5.25% sodium hypochlorite (31.6 g, 22.3 mmol) was adjusted to pH 7.1 with approximately 2 mL of 3 N HCl. Triethyl phosphonoacetate (1.0 g, 4.5 mmol) was added dropwise at ice-bath temperature with vigorous stirring. After complete addition, the ice-bath was removed, and stirring was continued for an additional 5 min. The turbid solution was extracted with  $5 \times 5$  mL of hexane. The combined hexane extracts were dried ( $\text{MgSO}_4$ ), and the solvent was removed in vacuo at  $50^\circ\text{C}$  to give triethyl 2,2-dichlorophosphonoacetate **6a** (1.18 g, 90% yield) as colourless oil. **6a** (1.1 g, 3.8 mmol) was dissolved in EtOH (7.5 mL), and the resulting solution was cooled in an ice-bath. A solution of sodium sulfite (1.0 g, 7.7 mmol) in  $\text{H}_2\text{O}$  (30 mL) was added with stirring at a rate such that the temperature could be maintained below  $15^\circ\text{C}$  (15 min). During addition, the reaction mixture became turbid. After 20 min of further stirring at room temperature, it was extracted with chloroform ( $5 \times 10$  mL). The chloroform extracts were dried ( $\text{MgSO}_4$ ), and the solvent was removed in vacuo. The crude mixture was partitioned between hexane (20 mL) and 0.1 M  $\text{NaHCO}_3$  (85 mL). The bicarbonate fractions were combined and re-extracted with chloroform ( $6 \times 5$  mL). The chloroform extracts were dried ( $\text{MgSO}_4$ ), and the solvent was removed in vacuo to give pure **7a** (0.85 g, 88% yield) as colourless oil.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.52 ppm (d,  $^2J_{\text{P,H}} = 16.2$ , 1H, CHCl), 4.35–4.23 ppm (m, 6H,  $\text{CH}_2$ ), 1.43–1.31 ppm (m, 9H,  $\text{CH}_3$ ).

##### 4.2. Triethyl 2-bromophosphonoacetate (7b) [21]

Triethyl phosphonoacetate (1.2 g, 5.4 mmol) was added over 3 min to the freshly prepared, stirred sodium hypobromite solution cooled in an ice-salt bath. The temperature was maintained below  $10^\circ\text{C}$ . When addition was complete, the mixture was immediately extracted with chloroform ( $4 \times 10$  mL). The chloroform extracts were washed with water ( $2 \times 2$  mL) and dried ( $\text{MgSO}_4$ ), and the solvent was removed in vacuo. The residue was partitioned between hexane (40 mL) and  $\text{H}_2\text{O}$  ( $2 \times 0.5$  mL) and the hexane extracts were dried ( $\text{MgSO}_4$ ). Removal of the solvent in vacuo left pure triethyl 2,2-dibromophosphonoacetate **6b** (1.78 g, 87% yield) as colourless oil. To **6b** (2.0 g, 5.2 mmol) dissolved in EtOH (5 mL) was added with cooling (ice-bath) a solution of 1.12 g (5.0 mmol) of  $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$  in  $\text{H}_2\text{O}$  (10 mL). The temperature was maintained below  $10^\circ\text{C}$ . When addition was complete (20 min), the reaction mixture was stirred for an additional 5 min at room temperature and then extracted with  $\text{CHCl}_3$  ( $4 \times 10$  mL). The chloroform extracts were dried ( $\text{MgSO}_4$ ,

and the solvent was removed in vacuo. The desired product was isolated by partitioning the crude residue between hexane (20 mL) and  $\text{H}_2\text{O}$  ( $4 \times 5$  mL). The aqueous fractions were combined and re-extracted with chloroform ( $3 \times 10$  mL). The chloroform extracts were dried ( $\text{MgSO}_4$ ) and evaporated at reduced pressure to provide pure **7b** (1.28 g, 81% yield) as colourless oil.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.37 ppm (d,  $^2J_{\text{P,H}} = 14.0$ , 1H, CHBr), 4.33–4.24 ppm (m, 6H,  $\text{CH}_2$ ), 1.40–1.30 ppm (m, 9H,  $\text{CH}_3$ ).

##### 4.3. General procedure for the preparation of 2-halo diethylphosphonoacetic acids **3** [26]

To a solution of sodium hydroxide (0.84 g, 15.0 mmol) in water (15 mL) was added the triethyl 2-halo phosphonoacetate (10.0 mmol). The reaction mixture was stirred for 20 h at r.t. and then washed with  $\text{CH}_2\text{Cl}_2$  ( $2 \times 10$  mL), acidified to pH 2 with 10% HCl and extracted with AcOEt ( $3 \times 15$  mL). After drying with  $\text{MgSO}_4$ , the solvent was evaporated under reduced pressure to give the acids **3**.

###### 4.3.1. 2-Chloro diethylphosphonoacetic acid (**3a**)

1.96 g (85% yield), colourless oil.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.59 ppm (d,  $^2J_{\text{P,H}} = 17.2$ , 1H, CHCl), 4.40–4.20 ppm (m, 4H,  $\text{CH}_2$ ), 1.42–1.37 ppm (m, 6H,  $\text{CH}_3$ ).

###### 4.3.2. 2-Bromo diethylphosphonoacetic acid (**3b**)

2.26 g (82% yield), colourless oil.  $^1\text{H}$  NMR (300 MHz, DMSO):  $\delta$  5.01 ppm (d,  $^2J_{\text{P,H}} = 14.6$ , 1H, CHBr), 4.19–3.98 ppm (m, 4H,  $\text{CH}_2$ ), 1.29–1.20 ppm (m, 6H,  $\text{CH}_3$ ).

###### 4.3.3. 2-Fluoro diethylphosphonoacetic acid (**3c**)

1.67 g (78% yield), colourless oil.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  5.26 ppm (dd,  $^2J_{\text{P,H}} = 13.4$ ,  $^2J_{\text{F,H}} = 47.1$ , 1H, CHF), 4.39–4.26 ppm (m, 4H,  $\text{CH}_2$ ), 1.42–1.36 ppm (m, 6H,  $\text{CH}_3$ ).

###### 4.3.4. 5-Chloro-2-nitroacetophenone (**9b**) [22]

To a rapidly stirred solution fuming  $\text{HNO}_3$  (8.5 ml) and concentrated  $\text{H}_2\text{SO}_4$  (1.3 ml) at  $-20^\circ\text{C}$  was added portionwise 3-chloroacetophenone (2.5 g, 16.2 mmol) over 15 min. The reaction mixture was allowed to warm to  $-10^\circ\text{C}$  and stirred for 5 h at this temperature after which ice-water (40 ml) was added and the reaction mixture extracted twice with  $\text{CH}_2\text{Cl}_2$ . The organic layers were combined, washed five times with water, dried over  $\text{MgSO}_4$  and concentrated under reduced pressure. The residue was filtered through a pad of silica (eluting with  $\text{CH}_2\text{Cl}_2/\text{PE}$ , 4:1) to afford a pale green oil which was recrystallized from  $\text{Et}_2\text{O}/\text{PE}$  to give 5-chloro-2-nitroacetophenone (2.52 g, 78% yield), pale yellow crystals, mp  $63$ – $65^\circ\text{C}$ . IR (KBr):  $\nu$  1712, 1565, 1520, 1421, 1340, 1307, 1245, 1105, 863  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.09 ppm (d,  $J_o = 8.7$ , 1H, Ar-H), 7.57 ppm (d,  $J_o = 8.7$ , 1H, Ar-H), 7.39 ppm (s, 1H, Ar-H), 2.56 ppm (s, 3H,  $\text{OCH}_3$ ).

###### 4.3.5. 2-Amino-5-chloroacetophenone (**4b**) [27]

A mixture of 5-chloro-2-nitroacetophenone (2.2 g, 11.0 mmol),  $\text{PtO}_2$  (20 mg) and charcoal (200 mg) in EtOH (40 ml) was rapidly stirred at r.t. for 4.5 h under 1 atm. of hydrogen. The reaction mixture was filtered through a pad of celite (the residues washed with  $\text{CH}_2\text{Cl}_2$ ) and concentrated under reduced pressure. The residues was filtered through a pad of silica (eluting with  $\text{CH}_2\text{Cl}_2/\text{PE}$ , 4:1) to afford a pale green oil which was recrystallized from  $\text{Et}_2\text{O}/\text{PE}$  to give 2-amino-5-chloroacetophenone (1.35 g, 72% yield), pale green crystals, mp  $63$ – $64^\circ\text{C}$ . IR (KBr):  $\nu$  3455, 3320, 1657, 1616, 1570, 1546, 1475, 1361, 1222, 1158, 956, 823, 626  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.67 ppm (s, 1H, Ar-H), 7.22 ppm (d,  $J_o = 8.7$ , 1H, Ar-H), 6.62 ppm (d,  $J_o = 8.7$ , 1H, Ar-H), 6.28 ppm (s, 2H,  $\text{NH}_2$ ), 2.57 ppm (s, 3H,  $\text{OCH}_3$ ).

#### 4.3.6. 4,5-Dimethoxy-2-nitroacetophenone (9c) [23]

Fuming nitric acid (6.0 mL, 140 mmol) was added dropwise to a solution of 3,4-dimethoxyacetophenone (3.6 g, 20.0 mmol) in AcOH (14 mL) at 0 °C. The reaction mixture was maintained at 0 °C for 10 min and at room temperature for 20 min. Then it was poured into ice and the separated residue was filtered off and washed with a water solution of NaHCO<sub>3</sub> until pH 7 was reached. Recrystallization from EtOH–acetone afforded compound **9c** (2.34 g, 52% yield), yellow needles, mp 135–137 °C. IR (KBr):  $\nu$  1702, 1578, 1515, 1464, 1326, 1284, 1225, 1183, 1047, 883, 789 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  7.62 ppm (s, 1H, Ar–H), 6.76 ppm (s, 1H, Ar–H), 3.99 ppm (s, 6H, OCH<sub>3</sub>), 2.51 ppm (s, 3H, OCH<sub>3</sub>).

#### 4.3.7. 2-Amino-4,5-dimethoxyacetophenone (4c) [23]

A mixture of 4,5-dimethoxy-2-nitroacetophenone (2.0 g, 9.0 mmol), iron powder (5.0 g), AcOH (18 mL), water (25 mL), and AcOEt (5 mL) was stirred under reflux for 6 h. After completion of the reaction, the mixture was neutralized with NaHCO<sub>3</sub> until pH 7 and filtered off. The residue was washed on the filter with AcOEt (3 × 40 mL). The organic layer was separated and the water layer was extracted with AcOEt (3 × 40 mL). Combined extracts were dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure. The residue was recrystallized from AcOEt–hexane afforded compound **4c** (1.19 g, 68% yield), pale yellow solid, mp 106–108 °C. IR (KBr):  $\nu$  3414, 3299, 1629, 1591, 1540, 1511, 1468, 1454, 1406, 1255, 1210, 1164, 1057, 949, 845, 566 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  7.10 ppm (s, 1H, Ar–H), 6.26 ppm (s, 2H, NH<sub>2</sub>), 6.11 ppm (s, 1H, Ar–H), 3.88 ppm (s, 3H, OCH<sub>3</sub>), 3.85 ppm (s, 3H, OCH<sub>3</sub>), 2.53 ppm (s, 3H, OCH<sub>3</sub>).

#### 4.4. General procedure for the preparation of *N*-acyl-*o*-aminophenylketone **2a–r**

To a solution of 2-halo-2-diethylphosphonoacetic acids **3a–c** (2.3 mmol) in dichloromethane (2 mL) was added oxalyl chloride (0.40 mL, 4.7 mmol) at 25 °C under argon. The mixture was stirred overnight and then concentrated. The brown oil was diluted with dichloromethane (5 mL), followed by the addition of the corresponding *o*-aminophenylketones **4a–f** (2.3 mmol) in dichloromethane (4 mL). In the case of **4f**, it was dissolved with dichloromethane (30 mL), due to the low solubility. Pyridine (0.38 mL, 4.7 mmol) was slowly added using a ice-water bath over 10 min. After stirring for 4.5 h, the reaction mixture was quenched with 3 N HCl (2 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 × 10 mL). The organic layer was washed with brine (5 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to obtain the corresponding product which was further purified by column chromatography (silica gel).

#### 4.4.1. Diethyl (2-acetylphenylcarbamoyl)chloromethylphosphonate (2a)

0.75 g (94% yield), purified by column chromatography (dichloromethane:methanol = 90:1), yellowish solid, mp 93–96 °C. IR (KBr):  $\nu$  3432, 1682, 1643, 1612, 1595, 1525, 1470, 1451, 1368, 1344, 1269, 1207, 1159, 1038, 1018, 869, 532 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  12.52 ppm (s, 1H, NH), 8.72 ppm (d,  $J$  = 8.4 Hz, 1H, Ar–H), 7.93 ppm (d,  $J$  = 7.9 Hz, 1H, Ar–H), 7.57 ppm (t,  $J$  = 7.6 Hz, 1H, Ar–H), 7.19 ppm (t,  $J$  = 7.6 Hz, 1H, Ar–H), 4.57 ppm (d, <sup>2</sup> $J_{P,H}$  = 16.1 Hz, 1H, CHCl), 4.37–4.24 ppm (m, 4H, CH<sub>2</sub>), 2.68 ppm (s, 3H, COCH<sub>3</sub>), 1.39–1.34 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  202.3, 162.8 ppm (d, <sup>2</sup> $J_{P,C}$  = 1.7 Hz), 139.7, 135.0, 131.5, 123.4, 122.4, 120.7, 64.6 ppm (d, <sup>2</sup> $J_{P,C}$  = 6.8 Hz), 52.8 ppm (d, <sup>1</sup> $J_{P,C}$  = 142.3 Hz), 28.3, 16.3 ppm (d, <sup>3</sup> $J_{P,C}$  = 5.8 Hz). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  5.38 ppm (s, 1P). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>14</sub>H<sub>19</sub>ClNO<sub>5</sub>PNa: 370.0582; found: 370.0586.

#### 4.4.2. Diethyl (2-acetyl-4-chlorophenylcarbamoyl)chloromethylphosphonate (2b)

0.82 g (93% yield), purified by column chromatography (dichloromethane:methanol = 90:1), yellow oil. IR (KBr):  $\nu$  3475, 1681, 1664, 1577, 1512, 1397, 1361, 1304, 1258, 1163, 1020, 961, 733 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  12.40 ppm (s, 1H, NH), 8.72 ppm (d,  $J_o$  = 9.0 Hz, 1H, Ar–H), 7.88 ppm (d,  $J_m$  = 2.1 Hz, 1H, Ar–H), 7.54 ppm (dd,  $J_m$  = 2.1 Hz,  $J_o$  = 9.0 Hz, 1H, Ar–H), 4.57 ppm (d, <sup>2</sup> $J_{P,H}$  = 16.2 Hz, 1H, CHCl), 4.37–4.25 ppm (m, 4H, CH<sub>2</sub>), 2.68 ppm (s, 3H, COCH<sub>3</sub>), 1.40–1.35 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  201.2, 162.9 ppm (d, <sup>2</sup> $J_{P,C}$  = 1.6 Hz), 138.2, 134.6, 131.1, 128.4, 123.6, 122.2, 64.7 ppm (d, <sup>2</sup> $J_{P,C}$  = 6.8 Hz), 52.8 ppm (d, <sup>1</sup> $J_{P,C}$  = 142.3 Hz), 28.3, 16.3 ppm (d, <sup>3</sup> $J_{P,C}$  = 5.8 Hz). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  6.01 ppm (s, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>14</sub>H<sub>18</sub>Cl<sub>2</sub>NO<sub>5</sub>PNa: 404.0192; found: 404.0192.

#### 4.4.3. Diethyl (2-acetyl-4,5-dimethoxyphenylcarbamoyl)chloromethylphosphonate (2c)

0.86 g (92% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellowish solid, mp 83–85 °C. IR (KBr):  $\nu$  3436, 1674, 1658, 1583, 1519, 1451, 1312, 1259, 1165, 1046, 1019, 970, 948, 777 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  12.79 ppm (s, 1H, NH), 8.49 ppm (s, 1H, Ar–H), 7.30 ppm (s, 1H, Ar–H), 4.59 ppm (d, 1H, <sup>2</sup> $J_{P,H}$  = 16.1 Hz, CHCl), 4.36–4.28 ppm (m, 4H, CH<sub>2</sub>), 3.98 ppm (s, 3H, OCH<sub>3</sub>), 3.93 ppm (s, 3H, OCH<sub>3</sub>), 2.64 ppm (s, 3H, COCH<sub>3</sub>), 1.40–1.35 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  200.4, 162.6 ppm (d, <sup>2</sup> $J_{P,C}$  = 1.7 Hz), 154.1, 144.0, 136.2, 114.8, 113.3, 103.4, 64.6 ppm (d, <sup>2</sup> $J_{P,C}$  = 6.7 Hz), 56.1, 56.0, 52.7 ppm (d, <sup>1</sup> $J_{P,C}$  = 143.3 Hz), 28.1, 16.2 ppm (d, <sup>3</sup> $J_{P,C}$  = 5.8 Hz). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  11.71 ppm (s, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>16</sub>H<sub>23</sub>ClNO<sub>7</sub>PNa: 430.0793; found: 430.0792.

#### 4.4.4. Diethyl (2-acetophenone-phenylcarbamoyl)chloromethylphosphonate (2d)

0.66 g (70% yield), purified by column chromatography (dichloromethane:methanol = 110:1), yellow oil. IR (KBr):  $\nu$  3485, 3278, 1692, 1641, 1583, 1528, 1448, 1317, 1294, 1269, 1163, 1051, 979, 755, 704 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  11.59 ppm (s, 1H, NH), 8.61 ppm (d,  $J$  = 8.7 Hz, 1H, Ar–H), 7.73 ppm (d,  $J$  = 7.4 Hz, 2H, Ar–H), 7.63 ppm (s, 1H, Ar–H), 7.59 ppm (d,  $J$  = 7.5 Hz, 2H, Ar–H), 7.49 ppm (t,  $J$  = 7.4 Hz, 2H, Ar–H), 7.16 ppm (t,  $J$  = 7.5 Hz, 1H, Ar–H), 4.59 ppm (d, <sup>2</sup> $J_{P,H}$  = 16.1 Hz, 1H, CHCl), 4.36–4.24 ppm (m, 4H, CH<sub>2</sub>), 1.40–1.31 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  198.8, 162.5 ppm (d, <sup>2</sup> $J_{P,C}$  = 1.6 Hz), 139.1, 138.1, 134.0, 133.3, 132.5, 129.9 × 2, 128.2 × 2, 124.1, 123.1, 121.5, 64.6 ppm (d, <sup>2</sup> $J_{P,C}$  = 6.7 Hz), 52.6 ppm (d, <sup>1</sup> $J_{P,C}$  = 142.5 Hz), 16.2 ppm (d, <sup>3</sup> $J_{P,C}$  = 5.8 Hz). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  10.88 ppm (s, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>21</sub>ClNO<sub>5</sub>PNa: 432.0738; found: 432.0744.

#### 4.4.5. Diethyl (2-acetophenone-4-chlorophenylcarbamoyl)chloromethylphosphonate (2e)

0.92 g (90% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellow oil. IR (KBr):  $\nu$  3427, 3238, 1697, 1663, 1604, 1528, 1482, 1332, 1293, 1274, 1224, 1163, 1043, 955, 697, 534 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  11.39 ppm (s, 1H, NH), 8.57 ppm (d,  $J$  = 9.6 Hz, 1H, Ar–H), 7.74 ppm (d,  $J$  = 7.5 Hz, 2H, Ar–H), 7.64 ppm (t,  $J$  = 7.3 Hz, 1H, Ar–H), 7.50–7.55 ppm (m, 4H, Ar–H), 4.58 ppm (d, <sup>2</sup> $J_{P,H}$  = 16.1 Hz, 1H, CHCl), 4.35–4.23 ppm (m, 4H, CH<sub>2</sub>), 1.37–1.31 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  197.4, 162.5 ppm (d, <sup>2</sup> $J_{P,C}$  = 1.6 Hz), 137.5, 137.4, 133.6, 133.0, 132.4, 129.9 × 2, 128.5 × 2, 128.4, 125.6, 123.1, 64.7 ppm (d, <sup>2</sup> $J_{P,C}$  = 6.8 Hz), 52.5 ppm (d, <sup>1</sup> $J_{P,C}$  = 142.4 Hz), 16.2 ppm (d, <sup>3</sup> $J_{P,C}$  = 5.8 Hz). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  6.04 ppm (s, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>20</sub>Cl<sub>2</sub>NO<sub>5</sub>PNa: 466.0348; found: 466.0341.

#### 4.4.6. Diethyl (2-acetophenone-4-nitrophenylcarbamoyl)chloromethylphosphonate (2f)

0.92 g (88% yield), purified by column chromatography (dichloromethane:methanol = 110:1), yellow oil. IR (KBr):  $\nu$  3471, 3253, 1704, 1648, 1616, 1582, 1541, 1511, 1446, 1413, 1346, 1262, 1157, 1022, 972, 732  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  11.84 ppm (s, 1H, NH), 8.88 ppm (d,  $J_o = 9.2$  Hz, 1H, Ar–H), 8.51 ppm (s, 1H, Ar–H), 8.45 ppm (dd,  $J_m = 2.0$  Hz,  $J_o = 9.2$  Hz, 1H, Ar–H), 7.77–7.67 ppm (m, 3H, Ar–H), 7.56 ppm (t,  $J = 7.5$  Hz, 2H, Ar–H), 4.64 ppm (d,  $^2J_{\text{P,H}} = 16.4$  Hz, 1H, CHCl), 4.37–4.26 ppm (m, 4H,  $\text{CH}_2$ ), 1.39–1.33 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  196.9, 163.0 ppm (d,  $^2J_{\text{P,C}} = 1.6$  Hz), 144.2, 141.9, 136.7, 133.4, 129.8  $\times$  2, 128.6  $\times$  2, 128.5, 128.2, 123.5, 121.5, 64.8 ppm (d,  $^2J_{\text{P,C}} = 6.9$  Hz), 52.5 ppm (d,  $^1J_{\text{P,C}} = 141.8$  Hz), 16.1 ppm (d,  $^3J_{\text{P,C}} = 5.8$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  6.20 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{19}\text{H}_{20}\text{ClN}_2\text{O}_7\text{PNa}$ : 477.0589; found: 477.0585.

#### 4.4.7. Diethyl (2-acetylphenylcarbamoyl)bromomethylphosphonate (2g)

0.82 g (91% yield), purified by column chromatography (dichloromethane:methanol = 90:1), yellowish solid, mp 88–90 °C. IR (KBr):  $\nu$  3426, 1669, 1658, 1583, 1516, 1451, 1311, 1258, 1166, 1047, 1018, 968, 948, 775  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  12.43 ppm (s, 1H, NH), 8.71 ppm (d,  $J = 8.4$  Hz, 1H, Ar–H), 7.93 ppm (d,  $J = 7.9$  Hz, 1H, Ar–H), 7.58 ppm (t,  $J = 7.4$  Hz, 1H, Ar–H), 7.19 ppm (t,  $J = 7.5$  Hz, 1H, Ar–H), 4.44 ppm (d,  $^2J_{\text{P,H}} = 14.5$  Hz, 1H, CHBr), 4.38–4.25 ppm (m, 4H,  $\text{CH}_2$ ), 2.69 ppm (s, 3H,  $\text{COCH}_3$ ), 1.40–1.35 ppm (t, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  202.2, 162.9, 139.9, 135.0 ppm (d,  $J = 2.8$  Hz), 131.5 ppm (d,  $J = 2.8$  Hz), 123.4 ppm (d,  $J = 1.8$  Hz), 122.5 ppm (d,  $J = 3.2$  Hz), 120.82, 64.76 ppm (d,  $^2J_{\text{P,C}} = 6.8$  Hz), 39.8 ppm (d,  $^1J_{\text{P,C}} = 140.0$  Hz), 28.4 ppm (d,  $J = 3.6$  Hz), 16.3 ppm (d,  $^3J_{\text{P,C}} = 5.6$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  13.39 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{14}\text{H}_{19}\text{BrNO}_5\text{PNa}$ : 414.0076; found: 414.0081.

#### 4.4.8. Diethyl (2-acetyl-4-chlorophenylcarbamoyl)bromomethylphosphonate (2h)

0.90 g (92% yield), purified by column chromatography (dichloromethane:methanol = 90:1), yellowish oil. IR (KBr):  $\nu$  3425, 1695, 1652, 1608, 1512, 1397, 1317, 1239, 1167, 1126, 1108, 1051, 1020, 976, 945, 762  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  12.30 ppm (s, 1H, NH), 8.70 ppm (d,  $J_o = 9.0$  Hz, 1H, Ar–H), 7.87 ppm (s, 1H, Ar–H), 7.53 ppm (dd,  $J_m = 1.5$  Hz,  $J_o = 9.0$  Hz, 1H, Ar–H), 4.43 ppm (d,  $^2J_{\text{P,H}} = 14.6$  Hz, 1H, CHBr), 4.37–4.28 ppm (m, 4H,  $\text{CH}_2$ ), 2.68 ppm (s, 3H,  $\text{COCH}_3$ ), 1.40–1.35 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  201.1, 162.9 ppm (d,  $^2J_{\text{P,C}} = 1.7$  Hz), 138.4, 134.6, 131.1, 128.4, 123.6, 122.3, 64.8 ppm (d,  $^2J_{\text{P,C}} = 6.9$  Hz), 39.57 ppm (d,  $^1J_{\text{P,C}} = 139.8$  Hz), 28.4, 16.3 ppm (d,  $^3J_{\text{P,C}} = 5.8$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.74 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{14}\text{H}_{18}\text{ClBrNO}_5\text{PNa}$ : 447.9687; found: 447.9683.

#### 4.4.9. Diethyl (2-acetyl-4,5-dimethoxyphenylcarbamoyl)bromomethylphosphonate (2i)

0.95 g (91% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellowish solid, mp 128–130 °C. IR (KBr):  $\nu$  3426, 1679, 1641, 1612, 1593, 1524, 1469, 1452, 1367, 1343, 1269, 1206, 1159, 1038, 1018, 980,  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  12.69 ppm (s, 1H, NH), 8.47 ppm (s, 1H, Ar–H), 7.30 ppm (s, 1H, Ar–H), 4.45 ppm (d,  $^2J_{\text{P,H}} = 14.6$  Hz, 1H, CHBr), 4.38–4.26 ppm (m, 4H,  $\text{CH}_2$ ), 3.98 ppm (s, 3H,  $\text{OCH}_3$ ), 3.92 ppm (s, 3H,  $\text{OCH}_3$ ), 2.63 ppm (s, 3H,  $\text{COCH}_3$ ), 1.39–1.35 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  200.4, 162.8 ppm (d,  $^2J_{\text{P,C}} = 1.6$  Hz), 154.3, 144.1, 136.6, 114.9, 113.4, 103.5, 64.7 ppm (d,  $^2J_{\text{P,C}} = 6.8$  Hz), 56.3, 56.2, 39.9 ppm (d,  $^1J_{\text{P,C}} = 140.8$  Hz), 28.2, 16.3 ppm (d,

$^3J_{\text{P,C}} = 5.9$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.50 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{16}\text{H}_{23}\text{BrNO}_7\text{PNa}$ : 474.0288; found: 474.0280.

#### 4.4.10. Diethyl (2-acetophenone-phenylcarbamoyl)bromomethylphosphonate (2j)

0.68 g (65% yield), purified by column chromatography (dichloromethane:methanol = 110:1), yellow oil. IR (KBr):  $\nu$  3480, 3274, 1693, 1642, 1601, 1583, 1528, 1447, 1317, 1294, 1267, 1162, 1051, 966, 755, 703  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  11.44 ppm (s, 1H, NH), 8.57 ppm (d,  $J = 8.5$  Hz, 1H, Ar–H), 7.74 ppm (d,  $J = 7.6$  Hz, 2H, Ar–H), 7.60–7.57 ppm (m, 3H, Ar–H), 7.49 ppm (t,  $J = 7.4$  Hz, 2H, Ar–H), 7.16 ppm (t,  $J = 7.6$  Hz, 1H, Ar–H), 4.45 ppm (d,  $^2J_{\text{P,H}} = 14.5$  Hz, 1H, CHBr), 4.36–4.24 ppm (m, 4H,  $\text{CH}_2$ ), 1.40–1.31 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  198.6, 162.5 ppm (d,  $^2J_{\text{P,C}} = 1.6$  Hz), 139.2, 138.1, 133.9, 133.2, 132.6, 129.9  $\times$  2, 128.2  $\times$  2, 124.3, 123.1, 121.6, 64.7 ppm (d,  $^2J_{\text{P,C}} = 6.8$  Hz), 39.4 ppm (d,  $^1J_{\text{P,C}} = 140.0$  Hz), 16.2 ppm (d,  $^3J_{\text{P,C}} = 5.8$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  10.39 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{19}\text{H}_{21}\text{BrNO}_5\text{PNa}$ : 476.0233; found: 476.0225.

#### 4.4.11. Diethyl (2-acetophenone-4-chlorophenylcarbamoyl)bromomethylphosphonate (2k)

0.98 g (87% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellow oil. IR (KBr):  $\nu$  3432, 3235, 1692, 1662, 1604, 1528, 1481, 1331, 1293, 1274, 1244, 1223, 1163, 1039, 975, 953, 696, 528  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  11.24 ppm (s, 1H, NH), 8.54 ppm (d,  $J = 9.6$  Hz, 1H, Ar–H), 7.75 ppm (d,  $J = 7.6$  Hz, 2H, Ar–H), 7.65 ppm (t,  $J = 7.1$  Hz, 1H, Ar–H), 7.56–7.50 ppm (m, 4H, Ar–H), 4.44 ppm (d,  $^2J_{\text{P,H}} = 14.5$  Hz, 1H, CHBr), 4.35–4.24 ppm (m, 4H,  $\text{CH}_2$ ), 1.37–1.31 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  197.2, 162.5 ppm (d,  $^2J_{\text{P,C}} = 1.4$  Hz), 137.6, 137.3, 133.5, 133.1, 132.3, 129.9  $\times$  2, 128.5  $\times$  2, 128.4, 125.8, 123.1, 64.8 ppm (d,  $^2J_{\text{P,C}} = 6.8$  Hz), 39.3 ppm (d,  $^1J_{\text{P,C}} = 140.0$  Hz), 16.2 ppm (d,  $^3J_{\text{P,C}} = 5.8$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.83 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{19}\text{H}_{20}\text{ClBrNO}_5\text{PNa}$ : 509.9843; found: 509.9842.

#### 4.4.12. Diethyl (2-acetophenone-4-nitrophenylcarbamoyl)bromomethylphosphonate (2l)

0.98 g (85% yield), purified by column chromatography (dichloromethane:methanol = 110:1), yellow oil. IR (KBr):  $\nu$  3472, 3253, 1702, 1650, 1617, 1582, 1541, 1510, 1412, 1346, 1262, 1157, 1021, 971, 732  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  11.68 ppm (s, 1H, NH), 8.84 ppm (d,  $J = 9.2$  Hz, 1H, Ar–H), 8.49 ppm (s, 1H, Ar–H), 8.44 ppm (d,  $J = 9.4$  Hz, 1H, Ar–H), 7.76 ppm (d,  $J = 7.7$  Hz, 2H, Ar–H), 7.69 ppm (t,  $J = 7.3$  Hz, 1H, Ar–H), 7.55 ppm (t,  $J = 7.4$  Hz, 2H, Ar–H), 4.48 ppm (d,  $^2J_{\text{P,H}} = 14.7$  Hz, 1H, CHBr), 4.34–4.25 ppm (m, 4H,  $\text{CH}_2$ ), 1.38–1.31 ppm (m, 6H,  $\text{CH}_3$ ).  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  197.0, 163.2 ppm (d,  $^2J_{\text{P,C}} = 1.6$  Hz), 144.6, 142.0, 136.9, 133.6, 130.0  $\times$  2, 128.8  $\times$  2, 128.7, 128.4, 123.7, 121.7, 65.0 ppm (d,  $^2J_{\text{P,C}} = 7.0$  Hz), 39.2 ppm (d,  $^1J_{\text{P,C}} = 139.4$  Hz), 16.3 ppm (d,  $^3J_{\text{P,C}} = 6.1$  Hz).  $^{31}\text{P}$  NMR (121 MHz,  $\text{CDCl}_3$ ):  $\delta$  3.61 ppm (s, 1P). HRMS-ESI:  $m/z$   $[\text{M}+\text{Na}]^+$  calcd for  $\text{C}_{19}\text{H}_{20}\text{BrN}_2\text{O}_7\text{PNa}$ : 521.0084; found: 521.0079.

#### 4.4.13. Diethyl (2-acetylphenylcarbamoyl)fluoromethylphosphonate (2m)

0.71 g (92% yield), purified by column chromatography (dichloromethane:methanol = 90:1), white solid, mp 126–129 °C. IR (KBr):  $\nu$  3441, 1688, 1655, 1584, 1526, 1451, 1362, 1314, 1258, 1164, 1044, 1017, 981, 959, 766  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  12.51 ppm (s, 1H, NH), 8.77 ppm (d,  $J = 8.4$  Hz, 1H, Ar–H), 7.94 ppm (d,  $J = 7.9$  Hz, 1H, Ar–H), 7.59 ppm (t,  $J = 7.6$  Hz, 1H, Ar–H),

7.21 ppm (t,  $J = 7.6$  Hz, 1H, Ar–H), 5.27 ppm (dd,  $^2J_{P,H} = 11.4$  Hz,  $^2J_{F,H} = 46.5$  Hz, 1H, CHF), 4.37–4.24 ppm (m, 4H, CH<sub>2</sub>), 2.68 ppm (s, 3H, COCH<sub>3</sub>), 1.41–1.35 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  202.4, 163.5 ppm (dd,  $^2J_{P,C} = 1.0$  Hz,  $^2J_{F,C} = 16.8$  Hz), 139.3, 135.0 ppm (d,  $J = 3.3$  Hz), 131.6 ppm (d,  $J = 3.2$  Hz), 123.5 ppm (d,  $J = 2.5$  Hz), 122.4, 120.9, 87.0 ppm (dd,  $^1J_{P,C} = 158.8$  Hz,  $^1J_{F,C} = 201.2$  Hz), 64.2 ppm (d,  $J_1 = 2.5$  Hz,  $J_2 = 6.3$  Hz), 28.3 ppm (d,  $J = 4.6$  Hz), 16.3 ppm (dd,  $J_1 = 1.5$  Hz,  $J_2 = 5.8$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –204.1 ppm (dd,  $^2J_{H,F} = 46.5$  Hz,  $^2J_{P,F} = 71.9$  Hz, 1F, CHF). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  9.79 ppm (d,  $J_F = 71.9$  Hz, 1P). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>14</sub>H<sub>19</sub>FNO<sub>5</sub>PNa: 354.0877; found: 354.0874.

#### 4.4.14. Diethyl (2-acetyl-4-chlorophenylcarbamoyl)fluoromethylphosphonate (2n)

0.79 g (92% yield), purified by column chromatography (dichloromethane:methanol = 90:1), white solid, mp 125–127 °C. IR (KBr):  $\nu$  3444, 1689, 1659, 1577, 1513, 1397, 1310, 1290, 1257, 1226, 1166, 1112, 1069, 1043, 1013, 973, 954, 761 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  12.40 ppm (s, 1H, NH), 8.78 ppm (d,  $J_o = 9.0$  Hz, 1H, Ar–H), 7.90 ppm (d,  $J_m = 2.0$  Hz, 1H, Ar–H), 7.56 ppm (dd,  $J_m = 2.0$  Hz,  $J_o = 9.0$  Hz, 1H, Ar–H), 5.28 ppm (dd,  $^2J_{P,H} = 11.5$  Hz,  $^2J_{F,H} = 46.8$  Hz, 1H, CHF), 4.36–4.28 ppm (m, 4H, CH<sub>2</sub>), 2.69 ppm (s, 3H, COCH<sub>3</sub>), 1.43–1.37 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  201.3, 163.5 ppm (d,  $^2J_{F,C} = 16.9$  Hz), 137.7, 134.6, 131.1, 128.4, 123.4, 122.3, 86.8 ppm (dd,  $^1J_{P,C} = 158.9$  Hz,  $^1J_{F,C} = 201.0$  Hz), 64.3 ppm (dd,  $J_1 = 2.2$  Hz,  $J_2 = 6.4$  Hz), 28.3, 16.3 ppm (dd,  $J_1 = 1.2$  Hz,  $J_2 = 5.8$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –205.5 ppm (dd,  $^2J_{H,F} = 46.8$  Hz,  $^2J_{P,F} = 71.9$  Hz, 1F, CHF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>14</sub>H<sub>18</sub>ClFNO<sub>5</sub>PNa: 388.0487; found: 388.0489.

#### 4.4.15. Diethyl (2-acetyl-4,5-dimethoxyphenylcarbamoyl)fluoromethylphosphonate (2o)

0.82 g (90% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellowish solid, mp 75–77 °C. IR (KBr):  $\nu$  3445, 1683, 1643, 1613, 1592, 1528, 1406, 1369, 1345, 1269, 1209, 1160, 1067, 1046, 1019 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  12.77 ppm (s, 1H, NH), 8.51 ppm (s, 1H, Ar–H), 7.30 ppm (s, 1H, Ar–H), 5.26 ppm (dd,  $^2J_{P,H} = 11.2$  Hz,  $^2J_{F,H} = 46.8$  Hz, 1H, CHF), 4.34–4.26 ppm (m, 4H, CH<sub>2</sub>), 3.98 ppm (s, 3H, OCH<sub>3</sub>), 3.92 ppm (s, 3H, OCH<sub>3</sub>), 2.63 ppm (s, 3H, COCH<sub>3</sub>), 1.41–1.34 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  200.7, 163.5 ppm (d,  $^2J_{F,C} = 17.4$  Hz), 154.3, 144.2, 136.0, 114.9, 113.4, 103.7, 86.9 ppm (dd,  $^1J_{P,C} = 159.2$  Hz,  $^1J_{F,C} = 201.6$  Hz), 64.3 ppm (dd,  $J_1 = 2.2$  Hz,  $J_2 = 6.6$  Hz), 56.3  $\times$  2, 28.3, 16.4 ppm (d,  $J = 5.6$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –204.6 ppm (dd,  $^2J_{H,F} = 46.8$  Hz,  $^2J_{P,F} = 70.5$  Hz, 1F, CHF). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  6.06 ppm (d,  $^2J_{F,P} = 70.5$  Hz, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>16</sub>H<sub>23</sub>FNO<sub>7</sub>PNa: 414.1088; found: 414.1085.

#### 4.4.16. Diethyl (2-acetophenone-phenylcarbamoyl)fluoromethylphosphonate (2p)

0.57 g (62% yield), purified by column chromatography (dichloromethane:methanol = 110:1), yellow oil. IR (KBr):  $\nu$  3481, 3286, 1696, 1641, 1584, 1523, 1450, 1296, 1267, 1163, 1018, 978, 923, 755, 733, 703, 644 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  11.60 ppm (s, 1H, NH), 8.67 ppm (d,  $J = 8.6$  Hz, 1H, Ar–H), 7.72 ppm (d,  $J = 7.5$  Hz, 2H, Ar–H), 7.61 ppm (t, 3H,  $J = 6.9$  Hz, Ar–H), 7.49 ppm (t, 2H,  $J = 7.5$  Hz, Ar–H), 7.17 ppm (t, 1H,  $J = 7.3$  Hz, Ar–H), 5.28 ppm (dd,  $^2J_{P,H} = 11.3$  Hz,  $^2J_{F,H} = 47.1$  Hz, 1H, CHF), 4.34–4.26 ppm (m, 4H, CH<sub>2</sub>), 1.39–1.33 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  199.1, 163.3 ppm (dd,  $^2J_{P,C} = 1.0$  Hz,  $^2J_{F,C} = 16.7$  Hz), 138.9, 138.2, 134.2, 133.6, 132.6, 129.9  $\times$  2, 128.3  $\times$  2, 123.9, 123.2, 121.7, 86.9 ppm (dd,  $^1J_{P,C} = 158.8$  Hz,  $^1J_{F,C} = 201.0$  Hz), 64.3 ppm (dd,  $J_1 = 2.0$  Hz,  $J_2 = 6.4$  Hz), 16.4 ppm

(d,  $J = 5.8$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –203.7 ppm (dd,  $^2J_{H,F} = 47.1$  Hz,  $^2J_{P,F} = 71.3$  Hz, 1F, CHF). <sup>31</sup>P NMR (121 MHz, CDCl<sub>3</sub>):  $\delta$  5.09 ppm (d,  $^2J_{F,P} = 71.3$  Hz, 1P). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>21</sub>FNO<sub>5</sub>PNa: 416.1034; found: 416.1036.

#### 4.4.17. Diethyl (2-acetophenone-4-chlorophenylcarbamoyl)fluoromethylphosphonate (2q)

0.89 g (89% yield), purified by column chromatography (dichloromethane:methanol = 100:1), yellow oil. IR (KBr):  $\nu$  3485, 3291, 1703, 1646, 1597, 1579, 1516, 1446, 1397, 1289, 1258, 1162, 1019, 975, 948, 702 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  11.42 ppm (s, 1H, NH), 8.67 ppm (d,  $J = 9.6$  Hz, 1H, Ar–H), 7.75 ppm (d,  $J = 7.4$  Hz, 2H, Ar–H), 7.67 ppm (t,  $J = 7.4$  Hz, 1H, Ar–H), 7.59–7.52 ppm (m, 4H, Ar–H), 5.29 ppm (dd,  $^2J_{P,H} = 11.4$  Hz,  $^2J_{F,H} = 46.8$  Hz, 1H, CHF), 4.35–4.25 ppm (m, 4H, CH<sub>2</sub>), 1.41–1.35 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  197.6, 163.2 ppm (d,  $^2J_{F,C} = 16.8$  Hz), 137.4, 137.2, 133.7, 133.0, 132.6, 129.8  $\times$  2, 128.4  $\times$  2, 128.3, 125.2, 123.1, 86.8 ppm (dd,  $^1J_{P,C} = 158.9$  Hz,  $^1J_{F,C} = 200.9$  Hz), 64.3 ppm (dd,  $J_1 = 2.6$  Hz,  $J_2 = 6.3$  Hz), 16.3 ppm (d,  $J = 5.7$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –204.6 ppm (dd,  $^2J_{H,F} = 46.8$  Hz,  $^2J_{P,F} = 70.5$  Hz, CHF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>20</sub>ClFNO<sub>5</sub>PNa: 450.0644; found: 450.0642.

#### 4.4.18. Diethyl (2-acetophenone-4-nitrophenylcarbamoyl)fluoromethylphosphonate (2r)

0.90 g (88% yield), purified by column chromatography (dichloromethane:methanol = 110:1), white solid, mp 101–103 °C. IR (KBr):  $\nu$  3443, 1706, 1641, 1617, 1581, 1543, 1510, 1351, 1254, 1156, 1100, 1021, cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  11.88 ppm (s, 1H, NH), 8.97 ppm (d,  $J_o = 9.2$  Hz, 1H, Ar–H), 8.55 ppm (d,  $J_m = 2.3$  Hz, 1H, Ar–H), 8.48 ppm (dd,  $J_m = 2.3$  Hz,  $J_o = 9.2$  Hz, 1H, Ar–H), 7.76 ppm (d,  $J = 7.6$  Hz, 2H, Ar–H), 7.70 ppm (d,  $J = 7.2$  Hz, 1H, Ar–H), 7.58 ppm (t,  $J = 7.5$  Hz, 2H, Ar–H), 5.34 ppm (dd,  $^2J_{P,H} = 11.8$  Hz,  $^2J_{F,H} = 46.5$  Hz, 1H, CHF), 4.37–4.27 ppm (m, 4H, CH<sub>2</sub>), 1.43–1.37 ppm (m, 6H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  197.4, 164.0 ppm (d,  $^2J_{F,C} = 16.4$  Hz), 144.1, 142.1, 136.9, 133.6, 129.9  $\times$  2, 128.9  $\times$  2, 128.8, 128.6, 123.3, 121.7, 86.9 ppm (dd,  $^1J_{P,C} = 158.6$  Hz,  $^1J_{F,C} = 200.7$  Hz), 64.5 ppm (dd,  $J_1 = 2.3$  Hz,  $J_2 = 6.4$  Hz), 16.3 ppm (d,  $J = 5.6$  Hz). <sup>19</sup>F NMR (282 MHz, CDCl<sub>3</sub>):  $\delta$  –204.4 ppm (dd,  $^2J_{H,F} = 46.5$  Hz,  $^2J_{P,F} = 70.8$  Hz, 1 F, CHF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>20</sub>FN<sub>2</sub>O<sub>7</sub>PNa: 461.0884; found: 461.0881.

#### 4.5. General procedure for the preparation of the halogenated 2-quinolones 1a–r

In a round-bottomed flask under argon, LiCl (0.13 g, 3 mmol) and N-acyl-*o*-aminophenylketones **2a–r** (1 mmol) was suspended in THF (4 mL). DBU (0.45 mL, 3 mmol) was slowly added for 10 min at r.t., then the mixture was stirred for 2 h. The volatiles were evaporated, the residue was taken up in ethyl acetate, and washed with 3 N HCl and satd NaHCO<sub>3</sub>. After drying (Na<sub>2</sub>SO<sub>4</sub>) and evaporation of the solvent, the corresponding compound was isolated from the residue by column chromatography or recrystallization.

##### 4.5.1. 3-Chloro-4-methyl-2-quinolone (1a) [28]

170.4 mg (88% yield), recrystallized from ethanol, white solid; mp 300–302 °C. IR (KBr):  $\nu$  3421, 1667, 1606, 1504, 1434, 1381, 746, 633 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  12.21 ppm (s, 1H, NH), 7.83 ppm (d,  $J = 8.1$  Hz, 1H, Ar–H), 7.56 ppm (t,  $J = 7.6$  Hz, 1H, Ar–H), 7.36 ppm (d,  $J = 8.1$  Hz, 1H, Ar–H), 7.28 ppm (t,  $J = 7.6$  Hz, 1H, Ar–H), 2.60 ppm (s, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  157.1, 144.3, 136.6, 130.6, 125.3, 125.1, 122.6, 119.2, 115.7, 16.3 ppm.

#### 4.5.2. 3,6-Dichloro-4-methyl-2-quinolone (1b) [16(c)]

216.7 mg (95% yield), recrystallized from ethanol, white solid; mp 272–274 °C. IR (KBr):  $\nu$  3445, 1648, 1603, 1487, 1409, 1146, 1098, 629  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.31 ppm (s, 1H, NH), 7.86 ppm (s, 1H, Ar-H), 7.60 ppm (d,  $J_o = 8.8$  Hz, 1H, Ar-H), 7.36 ppm (d,  $J_o = 8.7$  Hz, 1H, Ar-H), 2.59 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  156.8, 143.2, 135.4, 130.3, 126.5, 126.4, 124.4, 120.3, 117.4, 16.2 ppm.

#### 4.5.3. 3-Chloro-6,7-dimethoxy-4-methyl-2-quinolone (1c)

208.0 mg (82% yield), recrystallized from ethanol, white solid, mp 250–252 °C. IR (KBr):  $\nu$  3424, 1655, 1625, 1514, 1414, 1261, 1230, 1205, 1169, 1103, 1035  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.00 ppm (s, 1H, NH), 7.17 ppm (s, 1H, Ar-H), 6.88 ppm (s, 1H, Ar-H), 3.84 ppm (s, 3H, OCH<sub>3</sub>), 3.82 ppm (s, 3H, OCH<sub>3</sub>), 2.57 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.2, 152.1, 145.4, 144.0, 132.5, 122.5, 112.5, 106.6, 98.1, 56.2, 55.9, 16.7 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>12</sub>H<sub>12</sub>ClNO<sub>3</sub>Na: 276.0398; found: 276.0401.

#### 4.5.4. 3-Chloro-4-phenyl-2-quinolinone (1d)

240.4 mg (94% yield), recrystallized from ethanol, white solid, mp 191–193 °C. IR (KBr):  $\nu$  3423, 1658, 1611, 1597, 1437, 1072, 750, 631  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.45 ppm (s, 1H, NH), 7.61–7.53 ppm (m, 4H, Ar-H), 7.42 ppm (d,  $J = 8.1$  Hz, 1H, Ar-H), 7.36 ppm (s, 1H, Ar-H), 7.34 ppm (s, 1H, Ar-H), 7.14 ppm (t,  $J = 7.7$  Hz, 1H, Ar-H), 6.95 ppm (d,  $J = 8.0$  Hz, 1H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  162.4, 153.1, 142.2, 139.8, 135.7, 133.8  $\times$  2, 133.7, 133.6  $\times$  2, 131.6, 129.8, 127.5, 124.5, 120.6 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>10</sub>ClNO<sub>3</sub>Na: 278.0343; found: 278.0345.

#### 4.5.5. 3,6-Dichloro-4-phenyl-2-quinolone (1e) [16(d)]

275.6 mg (95% yield), recrystallized from ethanol, white solid, mp 299–301 °C. IR (KBr):  $\nu$  3420, 1656, 1480, 1407, 1070, 706, 642  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.56 ppm (s, 1H, NH), 7.58–7.50 ppm (m, 4H, Ar-H), 7.39 ppm (d,  $J = 8.8$  Hz, 1H, Ar-H), 7.31 ppm (d,  $J = 6.8$  Hz, 2H, Ar-H), 6.80 ppm (d,  $J = 1.4$  Hz, 1H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.1, 146.8, 136.0, 134.2, 130.4, 128.9, 128.9  $\times$  2, 128.5  $\times$  2, 126.4, 126.3, 125.1, 120.7, 117.6 ppm.

#### 4.5.6. 3-Chloro-6-nitro-4-phenyl-2-quinolone (1f)

288.7 mg (96% yield), recrystallized from ethanol, white solid, mp 221–223 °C. IR (KBr):  $\nu$  3423, 1655, 1606, 1529, 1486, 1334, 1066, 652  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  13.01 ppm (s, 1H, NH), 8.40 ppm (dd,  $J_m = 2.2$  Hz,  $J_o = 9.0$  Hz, 1H, Ar-H), 7.77 ppm (d,  $J_m = 2.1$  Hz, 1H, Ar-H), 7.67–7.56 ppm (m, 4H, Ar-H), 7.43 ppm (d,  $J = 6.7$  Hz, 2H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.5, 147.5, 141.8, 141.5, 133.7, 129.4, 129.1  $\times$  2, 128.7  $\times$  2, 127.1, 125.2, 122.5, 119.3, 116.9 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>9</sub>ClN<sub>2</sub>O<sub>3</sub>Na: 323.0194; found: 323.0197.

#### 4.5.7. 3-Bromo-4-methyl-2-quinolone (1g) [13(b)]

202.4 mg (85% yield), recrystallized from ethanol, white solid, mp 268–269 °C. IR (KBr):  $\nu$  3440, 1649, 1602, 1502, 1432, 1004, 749, 618  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.16 ppm (s, 1H, NH), 7.82 ppm (d,  $J = 8.1$  Hz, 1H, Ar-H), 7.55 ppm (t,  $J = 7.7$  Hz, 1H, Ar-H), 7.34 ppm (d,  $J = 8.1$  Hz, 1H, Ar-H), 7.24 ppm (t,  $J = 7.7$  Hz, 1H, Ar-H), 2.63 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.6, 147.5, 137.3, 131.1, 125.8, 122.8, 119.6, 119.5, 115.9, 19.9 ppm (d,  $J = 1.6$  Hz).

#### 4.5.8. 3-Bromo-6-chloro-4-methyl-2-quinolone (1h) [29]

239.8 mg (88% yield), recrystallized from ethanol, white solid, mp 204–206 °C. IR (KBr):  $\nu$  3450, 1648, 1602, 1486, 1142, 1097, 625  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.29 ppm (s, 1H, NH),

7.87 ppm (s, 1H, Ar-H), 7.60 ppm (dd,  $J_m = 1.6$  Hz,  $J_o = 8.7$  Hz, 1H, Ar-H), 7.35 ppm (d,  $J_o = 8.8$  Hz, 1H, Ar-H), 2.63 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.0, 146.2, 135.8, 130.5, 126.4, 124.6, 120.6, 120.3, 117.4, 19.6 ppm.

#### 4.5.9. 3-Bromo-6,7-dimethoxy-4-methyl-2-quinolone (1i)

232.5 mg (78% yield), recrystallized from ethanol, white solid, mp 227–229 °C. IR (KBr):  $\nu$  3443, 1654, 1514, 1410, 1258, 1230, 1207, 1169  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  11.97 ppm (s, 1H, NH), 7.20 ppm (s, 1H, Ar-H), 6.88 ppm (s, 1H, Ar-H), 3.84 ppm (s, 3H, OCH<sub>3</sub>), 3.82 ppm (s, 3H, OCH<sub>3</sub>), 2.63 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  158.1, 152.9, 147.4, 146.0, 133.7, 117.0, 113.3, 107.4, 98.7, 56.8, 56.5, 20.8 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>12</sub>H<sub>12</sub>BrNO<sub>3</sub>Na: 319.9893; found: 319.9889.

#### 4.5.10. 3-Bromo-4-phenyl-2-quinolinone (1j) [17(a)]

78.0 mg (26% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 228–230 °C. IR (KBr):  $\nu$  3444, 1656, 1604, 1110, 1056, 758, 598  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  11–12 ppm (br.s, 1H, NH), 7.59–7.52 ppm (m, 4H, Ar-H), 7.36 ppm (s, 1H, Ar-H), 7.31 ppm (d,  $J = 7.6$  Hz, 2H, Ar-H), 7.14 ppm (s, 1H, Ar-H), 7.13 ppm (s, 1H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  160.3, 152.7, 137.3, 137.1, 130.8, 128.7  $\times$  2, 128.5  $\times$  2, 127.2, 123.0, 120.8, 117.9, 116.6 ppm.

#### 4.5.11. 3-Bromo-6-chloro-4-phenyl-2-quinolone (1k)

60.2 mg (18% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 218–220 °C. IR (KBr):  $\nu$  3442, 1665, 1056, 1028, 1008, 823, 761  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.55 ppm (s, 1H, NH), 7.62–7.53 ppm (m, 4H, Ar-H), 7.42 ppm (d,  $J = 8.8$  Hz, 1H, Ar-H), 7.32 ppm (d,  $J = 6.8$  Hz, 2H, Ar-H), 6.80 ppm (s, 1H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.4, 150.0, 136.5, 136.4, 130.7, 129.0  $\times$  2, 128.9, 128.2  $\times$  2, 126.2, 125.4, 120.9, 120.4, 117.6 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>9</sub>BrClNO<sub>3</sub>Na: 355.9448; found: 355.9444.

#### 4.5.12. 3-Bromo-6-nitro-4-phenyl-2-quinolone (1l)

65.6 mg (19% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 202–204 °C. IR (KBr):  $\nu$  3440, 1669, 1058, 1028, 823, 761, 624  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.97 ppm (s, 1H, NH), 8.39 ppm (dd,  $J_m = 2.2$  Hz,  $J_o = 9.0$  Hz, 1H, Ar-H), 7.74 ppm (d,  $J_m = 2.1$  Hz, 1H, Ar-H), 7.67–7.60 ppm (m, 3H, Ar-H), 7.56 ppm (d,  $J_o = 9.1$  Hz, 1H, Ar-H), 7.40 ppm (d,  $J = 6.5$  Hz, 2H, Ar-H).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  157.7, 150.7, 142.0, 141.7, 135.9, 129.2, 129.0  $\times$  2, 128.4  $\times$  2, 125.4, 122.7, 121.1, 119.4, 116.8 ppm. HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>9</sub>BrN<sub>2</sub>O<sub>3</sub>Na: 366.9689; found: 366.9682.

#### 4.5.13. 3-Fluoro-4-methyl-2-quinolone (1m)

70.9 mg (40% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 179–181 °C. IR (KBr):  $\nu$  3444, 1666, 1629, 1577, 1437, 1204, 749, 698  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.15 ppm (s, 1H, NH), 7.70 ppm (d,  $J = 7.3$  Hz, 1H, Ar-H), 7.46 ppm (d,  $J = 6.9$  Hz, 1H, Ar-H), 7.32 ppm (d,  $J = 7.8$  Hz, 1H, Ar-H), 7.25 ppm (d,  $J = 6.4$  Hz, 1H, Ar-H), 2.35 ppm (s, 3H, CH<sub>3</sub>).  $^{13}\text{C}$  NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  155.4 ppm (d,  $^2J_{F,C} = 28.2$  Hz), 148.6 ppm (d,  $^1J_{F,C} = 243.6$  Hz), 135.6, 129.6 ppm (d,  $J = 2.0$  Hz), 127.7 ppm (d,  $^2J_{F,C} = 12.8$  Hz), 125.1 ppm (d,  $J = 6.4$  Hz), 122.8, 119.3 ppm (d,  $J = 4.5$  Hz), 115.8, 10.1 ppm (d,  $^3J_{F,C} = 4.4$  Hz).  $^{19}\text{F}$  NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  –133.1 ppm (s, 1F, CF).

#### 4.5.14. 6-Chloro-3-fluoro-4-methyl-2-quinolone (1n)

80.4 mg (38% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 211–213 °C. IR (KBr):  $\nu$  3445, 1664, 1631, 1574, 1492, 1422, 1202, 645  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR



NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.29 ppm (s, 1H, NH), 7.76 ppm (s, 1H, Ar-H), 7.54 ppm (d,  $J_o = 8.8$  Hz, 1H, Ar-H), 7.33 ppm (d,  $J_o = 8.8$  Hz, 1H, Ar-H), 2.36 ppm (s,  $J = 2.6$  Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  154.8 ppm (d,  $^2J_{F,C} = 28.3$  Hz), 148.8 ppm (d,  $^1J_{F,C} = 245.2$  Hz), 134.0 ppm (d,  $J = 2.0$  Hz), 129.3 ppm (d,  $J = 2.5$  Hz), 126.8 ppm (d,  $^2J_{F,C} = 13.7$  Hz), 126.7, 124.0 ppm (d,  $J = 6.5$  Hz), 120.5 ppm (d,  $J = 5.1$  Hz), 117.3, 9.8 ppm (d,  $^3J_{F,C} = 4.2$  Hz). <sup>19</sup>F NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  -129.93 ppm (s, 1F, CF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>10</sub>H<sub>7</sub>ClFNONa: 234.0092; found: 234.0093.

#### 4.5.15. 3-Fluoro-6,7-dimethoxy-4-methyl-2-quinolone (1o)

71.8 mg (30% yield), purified by column chromatography (dichloromethane:methanol = 30:1), white solid, mp 266–268 °C. IR (KBr):  $\nu$  3422, 1657, 1623, 1516, 1423, 1271, 1215, 1169, cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  11.96 ppm (s, 1H, NH), 7.11 ppm (s, 1H, Ar-H), 6.89 ppm (s, 1H, Ar-H), 3.84 ppm (s, 3H, OCH<sub>3</sub>), 3.80 ppm (s, 3H, OCH<sub>3</sub>), 2.50 ppm (s, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  154.7 ppm (d,  $^2J_{F,C} = 28.1$  Hz), 150.8 ppm (d,  $J = 2.5$  Hz), 147.1 ppm (d,  $^1J_{F,C} = 239.1$  Hz), 145.3, 130.4, 127.6 ppm (d,  $^2J_{F,C} = 14.2$  Hz), 111.7 ppm (d,  $J = 4.2$  Hz), 105.9 ppm (d,  $J = 5.6$  Hz), 98.0, 55.9, 55.5, 10.1 ppm (d,  $^3J_{F,C} = 4.6$  Hz). <sup>19</sup>F NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  -136.6 ppm (s, 1F, CF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>12</sub>H<sub>12</sub>FNO<sub>3</sub>Na: 260.0693; found: 260.0686.

#### 4.5.16. 3-Fluoro-4-phenyl-2-quinolinone (1p)

217.7 mg (91% yield), recrystallized from ethanol, white solid, mp 254–256 °C. IR (KBr):  $\nu$  3444, 1669, 1617, 1568, 1283, 755, 701 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  12.45 ppm (s, 1H, NH), 7.59–7.50 ppm (m, 4H, Ar-H), 7.46–7.41 ppm (m, 3H, Ar-H), 7.21–7.14 ppm (m, 2H, Ar-H). <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  155.3 ppm (d,  $^2J_{F,C} = 28.0$  Hz), 147.0 ppm (d,  $^1J_{F,C} = 246.6$  Hz), 135.7 ppm (d,  $J = 2.0$  Hz), 131.5 ppm (d,  $^2J_{F,C} = 12.8$  Hz), 130.1, 129.5 ppm (d,  $J = 2.6$  Hz), 129.4, 129.3, 129.0, 128.8 × 2, 125.9 ppm (d,  $J = 6.5$  Hz), 122.6, 118.6 ppm (d,  $J = 2.8$  Hz), 115.7. <sup>19</sup>F NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  -130.9 ppm (s, 1F, CF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>10</sub>FNONa: 262.0639; found: 262.0643.

#### 4.5.17. 6-Chloro-3-fluoro-4-phenyl-2-quinolone (1q)

251.8 mg (92% yield), recrystallized from ethanol, white solid, mp 197–199 °C. IR (KBr):  $\nu$  3445, 1661, 1620, 1569, 1484, 1414, 1271, 1203, 700, 664 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  11–14 ppm (br.s, 1H, NH), 7.61–7.55 ppm (m, 4H, Ar-H), 7.47–7.42 ppm (m, 3H, Ar-H), 7.04 ppm (s, 1H, Ar-H). <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  155.1 ppm (d,  $^2J_{F,C} = 28.0$  Hz), 147.7 ppm (d,  $^1J_{F,C} = 249.0$  Hz), 134.5 ppm (d,  $J = 3.4$  Hz), 130.5 ppm (d,  $^2J_{F,C} = 13.8$  Hz), 129.5, 129.4 ppm (d,  $J = 2.6$  Hz), 129.3 × 3, 129.0 × 2, 126.6, 124.6 ppm (d,  $J = 6.5$  Hz), 120.1 ppm (d,  $J = 3.4$  Hz), 117.7 ppm. <sup>19</sup>F NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  -127.56 ppm (s, 1F, CF). HRMS-MALDI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>9</sub>ClFNONa: 296.0249; found: 296.0256.

#### 4.5.18. 3-Fluoro-6-nitro-4-phenyl-2-quinolone (1r)

264.3 mg (93% yield), recrystallized from ethanol, white solid, mp 188–190 °C. IR (KBr):  $\nu$  3443, 1666, 1629, 1575, 1535, 1487, 1380, 1336, 1272, 1206, 832, 731 cm<sup>-1</sup>. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  8.30 ppm (d,  $J = 8.9$  Hz, 1H, Ar-H), 7.95 ppm (s, 1H, Ar-H), 7.64–7.62 ppm (m, 3H, Ar-H), 7.54–7.49 ppm (m, 3H, Ar-H). <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  155.6 ppm (d,  $^2J_{F,C} = 28.0$  Hz), 148.0 (d,  $^1J_{F,C} = 250.6$  Hz), 142.0, 140.3 ppm (d,  $J = 1.7$  Hz), 131.0 ppm (d,  $^2J_{F,C} = 13.9$  Hz), 129.7, 129.5 × 2, 129.1 × 2, 129.1, 124.2 ppm (d,  $J = 2.2$  Hz), 121.9 ppm (d,  $J = 7.0$  Hz), 118.7 ppm (d,  $J = 3.8$  Hz), 117.0 ppm. <sup>19</sup>F NMR (282 MHz, DMSO- $d_6$ ):  $\delta$  -127.0 ppm (s, 1F, CF). HRMS-ESI:  $m/z$  [M+Na]<sup>+</sup> calcd for C<sub>15</sub>H<sub>9</sub>FN<sub>2</sub>O<sub>3</sub>Na: 307.0489; found: 307.0493.

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## Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.jfluchem.2010.01.008.

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